

# Application News

#### **Total Organic Carbon Analysis**

#### **Carbon Measurement of Metal Powder Battery Material**

## No. **072**

Various metal powders are used in battery electrode materials, depending on the type. Because batteries are used in large quantities in familiar products such as mobile phones, notebook computers, and automobiles, a high level of safety is necessary. Therefore, high purity is demanded in the metal materials used as materials.

Quick, simple measurement of the concentration of carbon contained in metal powders is possible by using a Shimadzu total organic carbon (TOC) solid sample system.

This article introduces an example of total carbon measurement of lithium cobalt oxide, which is widely used as a positive electrode material in lithium ion batteries, by using Shimadzu solid sample system consisting of a TOC-L<sub>CPH</sub> total organic carbon analyzer and SSM-5000A solid sample combustion unit.

M. Tanaka

#### Analysis Method

Additives

Approximately 100 mg of a commercial lithium cobalt oxide (Fig. 2) powder reagent was placed in the sample boat of the SSM-5000A and weighed, and total carbon (TC) was measured. Next, analysis samples were prepared by adding glucose as a carbonaceous substance to 100 mg to 200 mg of lithium cobalt oxide so as to obtain carbon concentrations of 5.0 %, 1.0 %, and 0.2 %, and the TC of the samples was measured.

For calibration of the analyzer, a calibration curve was prepared by TC measurement of a glucose powder reagent (carbon concentration: 40 %).

#### Table 1 Measurement Conditions

	rubic i measurement conditions
Analyzer	: TOC solid sample system
	(TOC-L <sub>CPH</sub> TOC analyzer + SSM-5000A solid sample combustion unit)
C 11.1	
Cell length	: Short cell

TC oxidation : Combustion catalytic oxidation (Combustion method temperature: 900 °C)

Measurement item : TC (Total carbon)

Calibration curve : One point calibration curve by glucose powder reagent (Carbon concentration: 40 %)

Sample : Lithium cobalt oxide (Lithium cobalt (III) oxide, LiCoO<sub>2</sub>) reagent

5 % addition, glucose powder; 1 % addition,
 5 %C glucose solution; 0.2 % addition, 0.5 %C glucose solution



Fig. 1 Lithium Ion Battery



Fig. 2 Lithium Cobalt Oxide

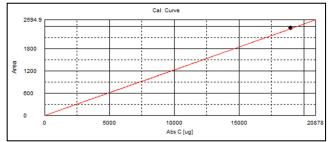
#### Analysis Results

Table 2 shows the results of measurements of the lithium cobalt oxide and the samples with the added carbonaceous substance. Fig. 3 shows the measurement charts. It can be understood that the samples with the added carbonaceous substance were measured with high accuracy.

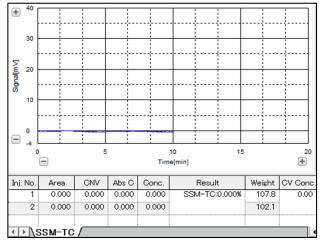
**Table 2 Measurement Results** 

Sample	TC Measurement Value (%C)
Lithium Cobalt Oxide	0
Lithium Cobalt Oxide + 0.2 % Glucose	0.209
Lithium Cobalt Oxide + 1.0 % Glucose	0.999
Lithium Cobalt Oxide + 5.0 % Glucose	5.02

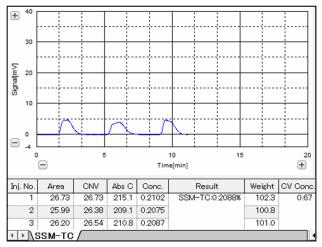
#### Analysis Data



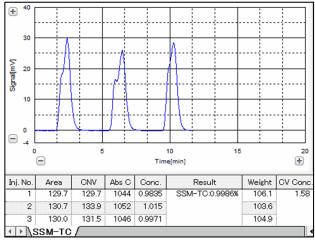
Calibration Curve



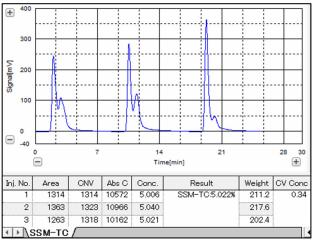
Sample: Lithium Cobalt Oxide



Sample: Lithium Cobalt Oxide + 0.2 % Glucose



Sample: Lithium Cobalt Oxide + 1.0 % Glucose



Sample: Lithium Cobalt Oxide + 5.0 % Glucose

Fig. 3 Measurement Data

First Edition: Dec. 2018



Shimadzu Corporation www.shimadzu.com/an/

#### For Research Use Only. Not for use in diagnostic procedures.

This publication may contain references to products that are not available in your country. Please contact us to check the availability of these products in your country.

The content of this publication shall not be reproduced, altered or sold for any commercial purpose without the written approval of Shimadzu. Shimadzu disclaims any proprietary interest in trademarks and trade names used in this publication other than its own. See <a href="http://www.shimadzu.com/about/trademarks/index.html">http://www.shimadzu.com/about/trademarks/index.html</a> for details.

The information contained herein is provided to you "as is" without warranty of any kind including without limitation warranties as to its accuracy or completeness. Shimadzu does not assume any responsibility or liability for any damage, whether direct or indirect, relating to the use of this publication. This publication is based upon the information available to Shimadzu on or before the date of publication, and subject to change without notice.

### **Related Products** Some products may be updated to newer models.



### **Related Solutions**

