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Analysis of Trichothecenes by Gas Chromatography-Chemical Ionization Mass Spectrometry

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Trichothecenes is one group of toxins produced by fungi (mycotoxins). Trichothecenes have been classified into four groups, A, B, C and D. Langseth and Rundberget^[1] reviewed some instrumental methods for the determination of trichothecenes in foodstuffs.

Among the instrumental methods used is gas chromatography with mass spectrometric detection (GC/MS). Here we report the analysis of eight trichothecenes by GC/MS based on the method described by Schollenberger et al.^[2] The literature described the use of GC/MS with chemical ionization.

In GC/MS analysis, the accuracy of quantitation and identification of a compound in a complex matrix will be improved if the compound's mass spectrum has ions that are relatively characteristic to the compound. The higher

the m/z values the more characteristic the ions are. The most characteristic ion of a compound would be the ion whose m/z value corresponds to the molecular weight of the compound. Meanwhile, the sensitivity of detection for the compound will be improved if such characteristic ions have relatively high abundances.

Chemical ionization mass spectrometry is very suitable for obtaining ions with high m/z of a compound, due to the "soft" nature of this ionization technique, that is, the energy involved in the ionization process is relatively less harsh than that involved in the more commonly used electron impact ionization mass spectrometry.

The eight trichothecenes analyzed belong to type A and type B trichothecenes, and are shown in Table 1 and Figure 1 below.

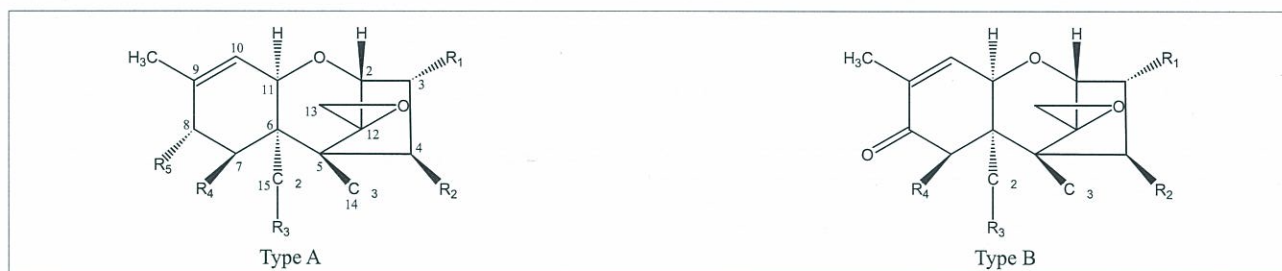


Figure 1. Structures of Type A and Type B trichothecenes.

Table 1. Trichothecenes analyzed in this study.

Name	Substituent at position					MW
	R ₁	R ₂	R ₃	R ₄	R ₅	
HT-2 toxin (HT-2)	OH	OH	OAc	H	OCOCH ₂ CH(CH ₃) ₂	424
T-2 toxin (T-2)	OH	OAc	OAc	H	OCOCH ₂ CH(CH ₃) ₂	466
Diacetoxyscirpenol (DAS)	OH	OAc	OAc	H	H	366
Deoxynivalenol (DON)	OH	H	OH	OH	-	294
3-Acetyl-deoxynivalenol (3-ADON)	OAc	H	OH	OH	-	338
15-Acetyl-deoxynivalenol (15-ADON)	OH	H	OAc	H	-	338
Nivalenol (NIV)	OH	OH	OH	OH	-	312
Fusarenon-X (FUS-X)	OH	OAc	OH	OH	-	354

- [1] W. Langseth, T. Rundberget, *Instrumental methods for determination of nonmacrocylic trichothecenes in cereals, foodstuffs and cultures*, *Journal of Chromatography A*, **815** (1998), 103-121
- [2] M. Schollenberger et al, *Determination of eight trichothecenes by gas chromatographymass spectrometry after sample clean-up by two-stage solid-phase extraction*, *Journal of Chromatography A*, **815** (1998), 123-132

Experimental

Trichothecenes standards and trifluoroacetic anhydride (TFAA) were purchased from Sigma-Aldrich (Singapore). The trichothecenes standards were derivatized by trifluoroacetylation according to the method described by Schollenberger et al.,^[2] and outlined in Figure 2.

The analysis was performed by using Shimadzu GCMS-QP2010nc in the Chemical Ionization (CI) mode. The analytical conditions are shown in Figure 3.

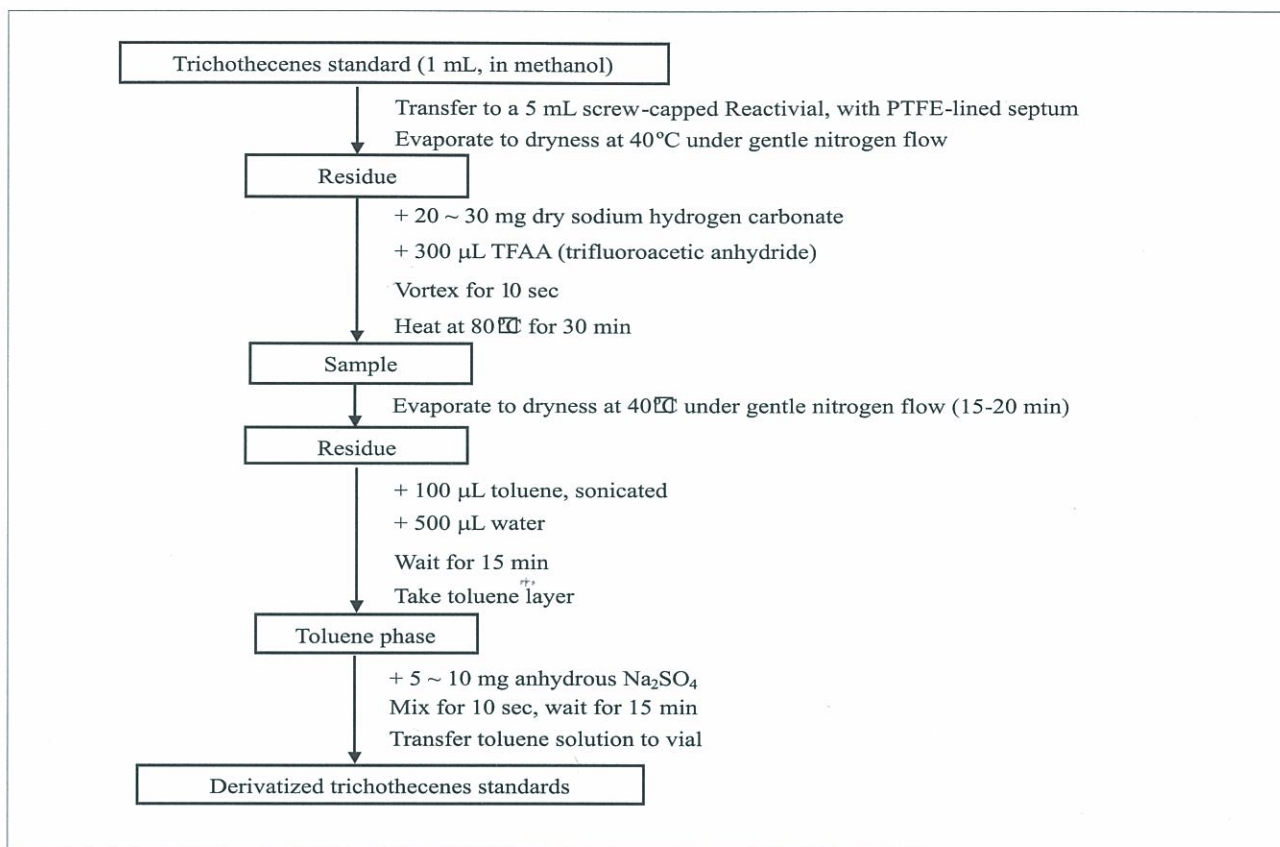


Figure 2. Derivatization steps of trichothecenes.

ANALYTICAL CONDITION			
Instrument: Shimadzu GCMS-QP2010nc + AOC-20i auto-injector			
GC Condition		MS Condition	
Injector temp.:	260 °C	Ionization mode:	Chemical Ionization
Injection mode:	Pulsed Splitless	Reagent gas:	Isobutane, 99.9%, 0.5bar
Sampling time:	1.5 min	Ion source:	200°C
High Pressure injection:	250 kPa (2min)	Interface temp:	290°C
Flow control mode:	Linear velocity	Solvent cut time:	3min
Linear velocity:	40 cm/s	Acquisition mode:	SCAN & SIM
Column flow:	1.17 ml/min	SCAN m/z range:	280-900 u
Column:	Rtx-5SilMS	SIM m/z:	Refer to NOTE in Table 2
Column temp. Program:	30m, 0.25mm ID, 0.25um df 90°C (2min) - 23°C/min - 275°C(2min) - 30°C/min - 290°C (15min)		

Figure 3. GCMS analysis conditions.

[2] M. Schollenberger et al, Determination of eight trichothecenes by gas chromatography-mass spectrometry after sample clean-up by two-stage solid-phase extraction, *Journal of Chromatography A*, **815** (1998), 123-132

Results

Trichothecenes Chromatogram & Mass Spectra

The total ion chromatogram of the 10 ppm trichothecenes mixture (after derivatization) obtained in Full scan mode was shown in Figure 3. The mass spectra of each compound are shown in Figure 5. Identification of the compounds was based on the comparison of mass spectra and elution order obtained with those reported in literature.^[2]

The results obtained by GCMS-QP2010 showed that for Type B trichothecenes analyzed, the molecular ion was the most abundant ion in the mass spectrum (see Table 2). Whereas for Type A trichothecenes analyzed, the molecular ion was the second most abundant ion in the mass spectrum.

Trichothecenes Calibration Curves

Figure 6 shows the SIM data and calibration curves for the trichothecenes derivatives.

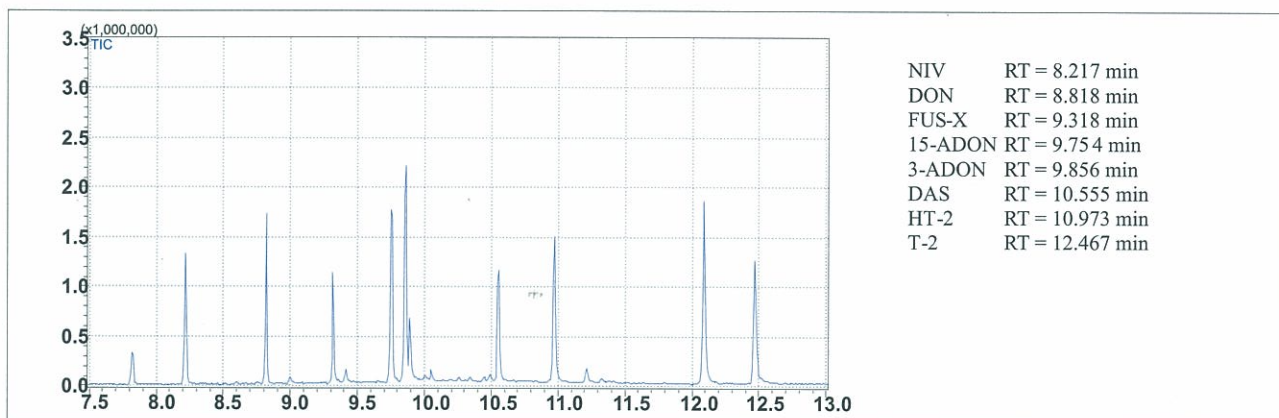


Figure 4. Total Ion Chromatogram (TIC) of 10 ppm trichothecenes standard solution (derivatized, analyzed in SCAN mode).

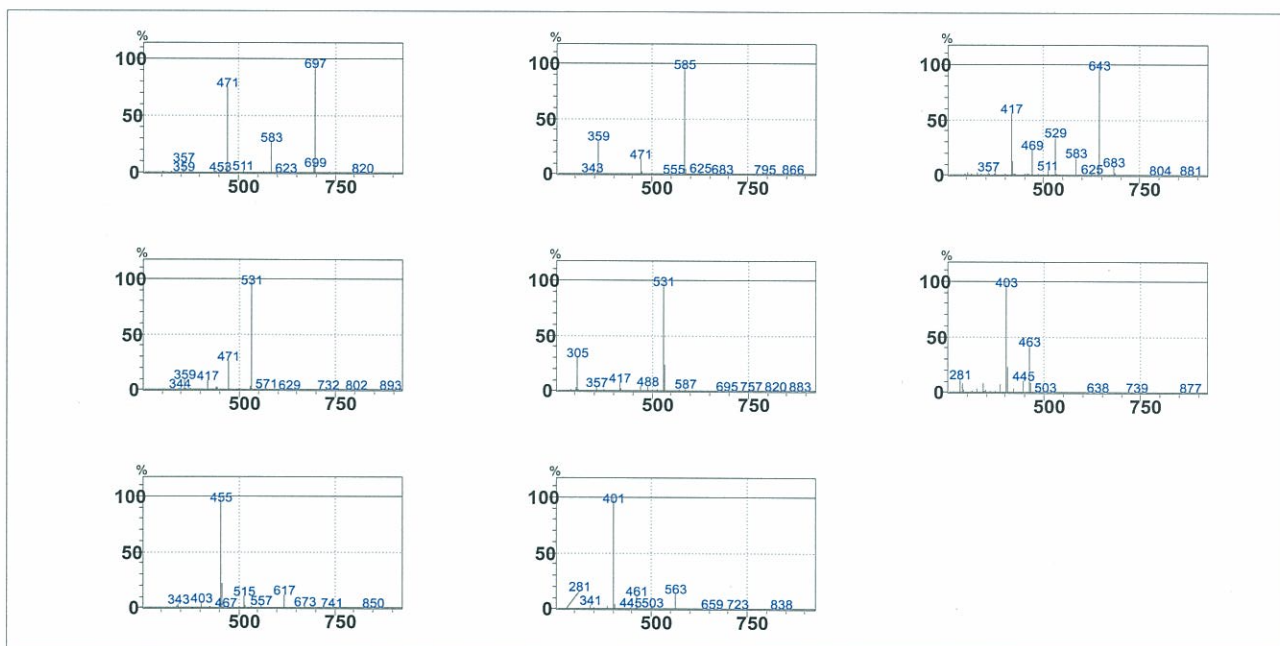


Figure 5. Chemical ionization mass spectra of the trichothecene derivatives.

Table 2. Mass spectral analysis results of trifluoroacetyl derivatives of eight trichothecenes (CI mode, isobutane as reagent gas).

Name	Observation				Literature ^[2]		
	Molecular ion ($M+H$) ⁺ (u)	Main fragment (u)	Secondary fragment (u)	Tertiary fragment (u)	Molecular ion ($M+H$) ⁺ (u)	Main fragment (u)	Secondary fragment (u)
NIV	697	697	471	583	697*	583	469
DON	585	585	359	471	585	585	479
FUS-X	643	643	417	529	643	529	469
15-ADON	531	531	471	417	531	471	417
3-ADON	531	531	305	417	531	531	471
DAS	463	403	463	445	463	403	289
HT-2	617	455	617	515	617	455	617
T-2	563	401	563	461	563	401	563

NOTE: main fragment and secondary fragment ions were used for the SIM analysis

*out of mass range of instrument; not reported

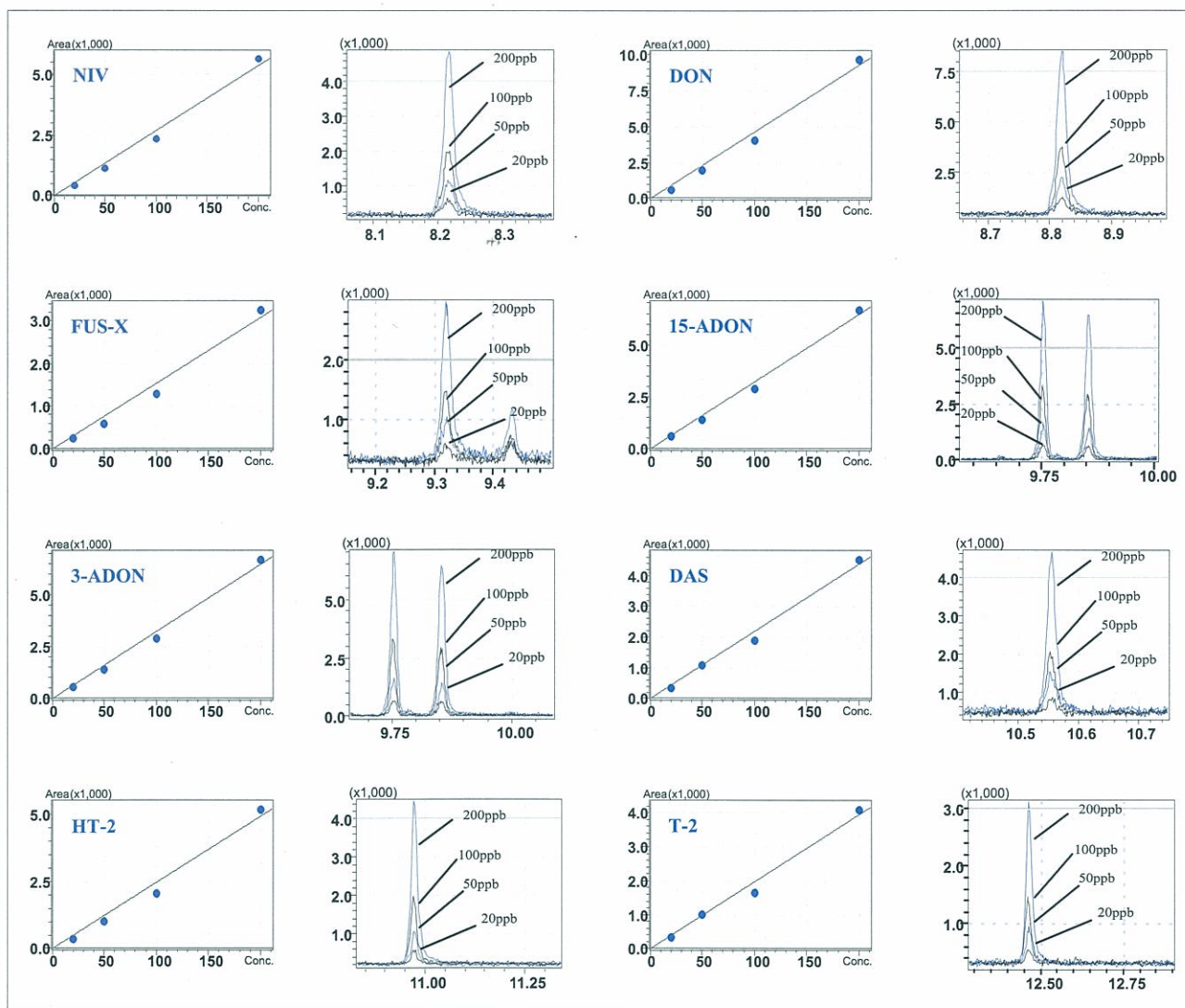


Figure 6. Calibration curves and SIM data for trichothecenes derivatives.

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