

Application News

Inductively Coupled Plasma Atomic Emission Spectrometry

Analysis of Plants – 2

■ Description

The multi-type ICPE-9000 ICP emission spectrometer was used to analyze several plant standards (tea leaf, olive, citrus leaf). Figs. 1 to 3 show the semi-quantitation values, and Table 1 shows the quantitation results. The results closely agree with the certified value for each element. Figs. 4 to 7 show the spectral profiles and calibration curves.

■ Sample

- Powdered tea leaf standard NIES No.7
- Olive standard BCR CRM-62
- Citrus Leaf SRM1572

■ Pretreatment

Perform heat-digestion using nitric acid, hydrochloric acid and a small amount of hydrofluoric acid. After letting the sample cool, adjust the volume appropriately, and use this as the analytical sample.

■ Analytical Conditions

Instrument	: ICPE-9000
Radio Frequency Power	: 1.2 (kW)
Plasma Gas	: 10 (L/min)
Auxiliary Gas	: 0.6 (L/min)
Carrier Gas	: 0.8 (L/min)
Sample Introduction	: Coaxial Nebulizer
Sample Aspiration	: 1.0 (mL/min)
Misting Chamber	: Cyclone Chamber
Attached Instruments	: Mini Torch
View Direction	: Axial/Radial

Table 1: Quantitation Results of Plant Standards ($\mu\text{g/g}$)

Element	Tea Leaf (NIES No. 7)		Olive (BCR CRM-62)		Citrus Leaf (NIST SRM1572)	
	Quantitation Value	Certified Value	Quantitation Value	Certified Value	Quantitation Value	Certified Value
Al	760	775 \pm 20	256	(260)	101	92 \pm 15
As	<0.4		<0.4		2.8	3.1
Ba	13.5		83		22.0	21
Ca	3280	3200 \pm 120	16800	(17500)	30800	31500 \pm 1000
Cd	0.03	0.03	0.11	0.1	0.03	0.03
Cr	0.17	(0.15)	1.03		0.71	0.8 \pm 0.2
Cu	7.0	7 \pm 0.3	45.3	46.6 \pm 1.8	15.7	16.5 \pm 1.0
Fe	95.4		284	(280)	84	90 \pm 10
K	18800	18600 \pm 700	3300	(3100)	17800	18200 \pm 600
Mg	1480	1530 \pm 60	1120	(1200)	5690	5800 \pm 300
Mn	706	700 \pm 25	55.7	57 \pm 2.4	21.8	23 \pm 2
Ni	6.5	6.5 \pm 0.3	1.05		0.52	0.6 \pm 0.3
P	3600	(3700)	1030	(1050)	1250	1300 \pm 200
Pb	0.8	0.8 \pm 0.03	24.3	25 \pm 1.5	12.0	13.3 \pm 2.4
Sr	3.70	3.7	35.2		98.0	100
V	0.06		0.72		0.22	
Zn	31.1	33 \pm 3	15.4	16 \pm 0.7	28.8	29 \pm 2

*Values in parentheses are reference values

Tea Leaf / 25						
1000 mg/L or greater						
1 mg/L or greater	Al 71 +	B 1.2	Ca 70	Fe 8.4	K 145	Mg 46
	Mn 25	Na 5.2	P 150	S 11	Si 1.6	
1 µg/L or greater	Ba 140	Cr 37	Cu 740	Dy 26	In 250	Li 3.3
	Ni 320	Rh 190	Sr 58	Th 220	Ti 21	Zn 550
Up to 1 µg/L						
Below detection limit µg/L	Ag < 14	As < 380	Au < 130	Be < 0.43	Bi < 470	Cd < 14
	Ce < 66	Co < 36	Er < 22	Eu < 3.1	Ga < 57	Gd < 18
	Ge < 430	Hf < 79	Hg < 10	Ho < 9.7	I < 1400	Ir < 880
	La < 13	Lu < 4.0	Mo < 70	Nb < 110	Nd < 44	Os < 510 +
	Pb < 220	Pd < 44	Pr < 46	Pt < 210	Re < 57	Ru < 100
	Sb < 290	Sc < 2.0	Se < 490	Sm < 29	Sn < 180	Ta < 71
	Tb < 20	Te < 540	Tl < 320	Tm < 8.6	U < 190	V < 37
	W < 510	Y < 2.3	Yb < 7.1	Zr < 4.1		

Figure 1: Semi-Quantitation Values (Tea Leaf Digest Solution)

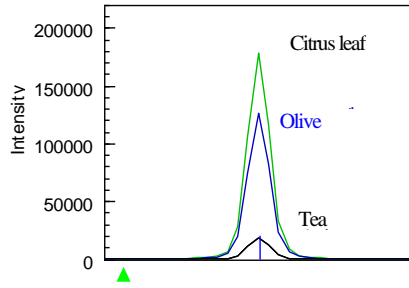
Olive / 20						
1000 mg/L or greater						
1m g/L or greater	Al 47 +	B 1.4	Ba 2.6	Ca 540	Cu 5.7	Fe 32
	K 74	Mg 43	Mn 2.7	Na 13	P 59	Pb 1.4
	S 14	Sr 1.6				
1 µg/L or greater	Cr 97	Li 7.5	Ni 78	Si 140	Th 30	Ti 200
	V 57	Y 4.3	Zn 350			
Up to 1 µg/L						
Below detection limit µg/L	Ag < 21	As < 560	Au < 170	Be < 0.91	Bi < 430	Cd < 20
	Ce < 110	Co < 43	Dy < 38	Er < 34	Eu < 5.5	Ga < 96
	Gd < 28	Ge < 270	Hf < 100	Hg < 15	Ho < 48	I < 2300
	In < 200	Ir < 1200	La < 23	Lu < 5.2	Mo < 95	Nb < 120
	Nd < 86	Os < 770 +	Pd < 66	Pr < 76	Pt < 1300	Re < 78
	Rh < 120	Ru < 310	Sb < 410	Sc < 3.5	Se < 710	Sm < 48
	Sn < 250	Ta < 110	Tb < 31	Te < 720	Tl < 510	Tm < 14
	U < 330	W < 770	Yb < 11	Zr < 6.1		

Figure 2: Semi-Quantitation Values (Olive Digest Solution)

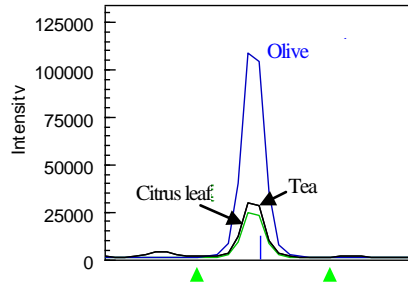
Citrus Leaf / 25						
1000 mg/L or greater						
1 mg/L or greater	Al 7.8 +	B 3.6	Ca 760	Cu 1.7	Fe 6.9	K 249
	Mg 160	Na 17	P 59	S 19	Si 1.0	Sr 3.5
1 µg/L or greater	Ba 410	Cr 55	Li 4.2	Mn 830	Pb 560	Rh 230
	Th 55	Ti 39	Y 5.9	Zn 460		
Up to 1 µg/L						
Below detection limit µg/L	Ag < 24	As < 700	Au < 75	Be < 1.1	Bi < 500	Cd < 25
	Ce < 130	Co < 59	Dy < 17	Er < 40	Eu < 6.4	Ga < 110
	Gd < 33	Ge < 350	Hf < 130	Hg < 18	Ho < 17	I < 2900
	In < 230	Ir < 1500	La < 27	Lu < 6.0	Mo < 120	Nb < 180
	Nd < 100	Ni < 60	Os < 750 +	Pd < 76	Pr < 87	Pt < 370
	Re < 88	Ru < 180	Sb < 510	Sc < 4.1	Se < 880	Sm < 56
	Sn < 320	Ta < 120	Tb < 36	Te < 870	Tl < 580	Tm < 16
	U < 380	V < 65	W < 960	Yb < 13	Zr < 7.4	

Figure 3: Semi-Quantitation Values (Citrus Leaf Digest Solution)

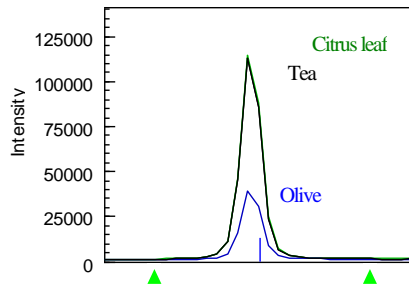
Ca 315.887 Best
Cond 2



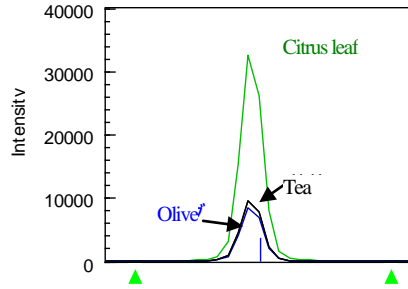
Fe 259.940
Cond 1



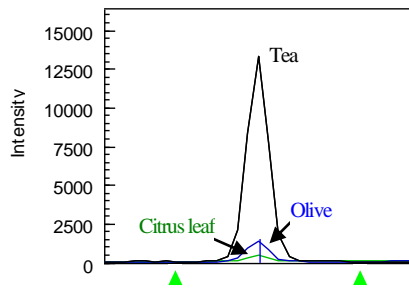
K 766.490
Cond 2



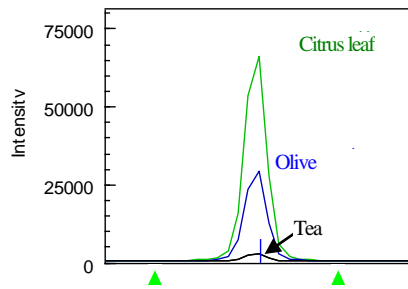
Mg 285.213 Best
Cond 2



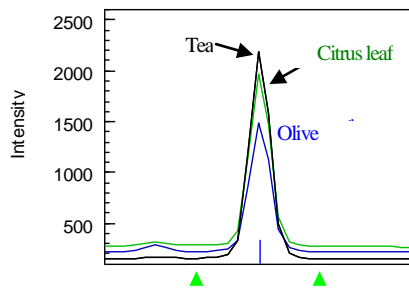
Mn 257.610 Best
Cond 2



Sr 421.552
Cond 2



Zn 206.200
Cond 1



Cu 324.754 Best
Cond 1

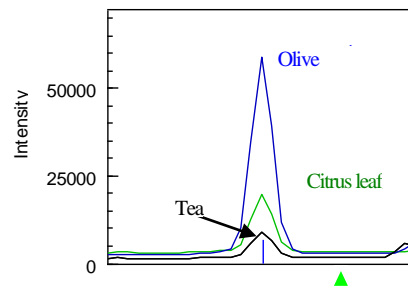
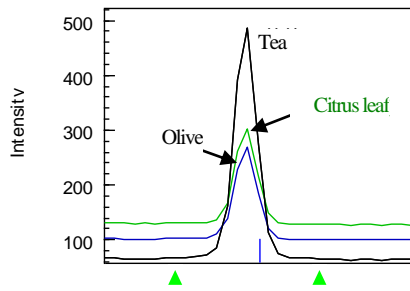
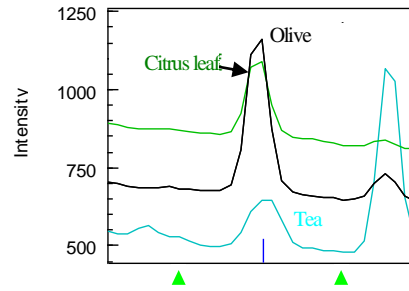


Figure 4: Spectral Profiles

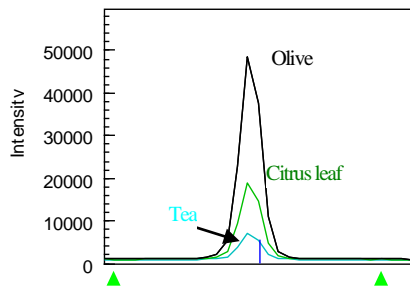
P 178.287
Cond 1



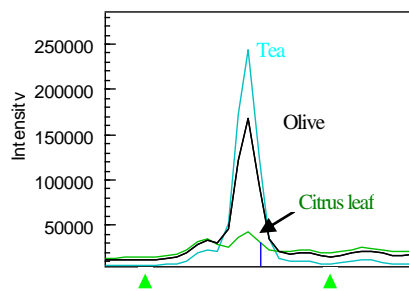
Cr 267.716 Best
Cond 1



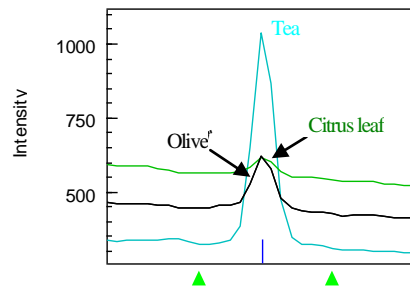
Ba 455.403 Best
Cond 2



Al 396.153
Cond 1



Ni 231.604 Best
Cond 1



Pb 220.353 Best
Cond 1

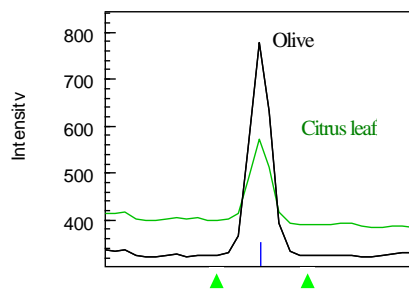
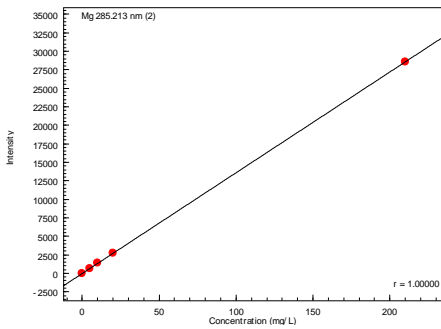
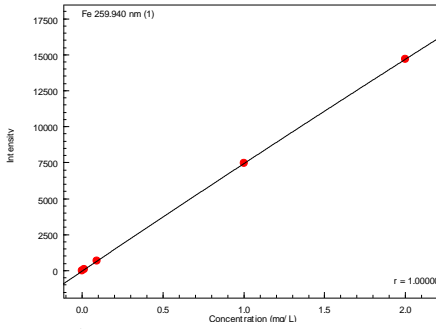


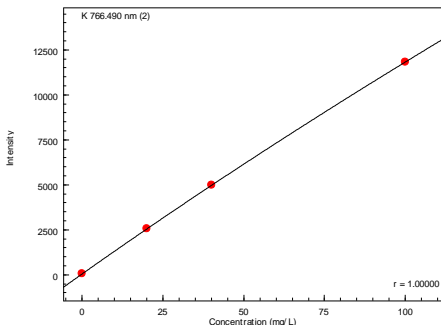
Figure 5: Spectral Profiles



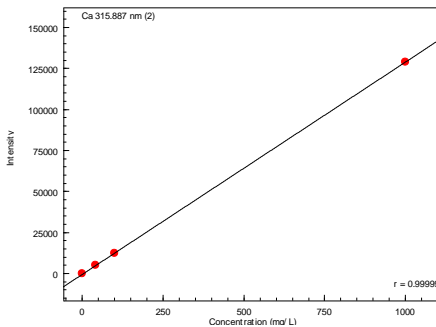
Equation: $\text{Conc.} = a \cdot I^3 + b \cdot I^2 + c \cdot I + d$
 Factor: $a = 0.0000000$ $c = 0.0073434$ **Weight: None**
 $b = 0.0000000$ $d = -0.1725735$ **Offset: None**



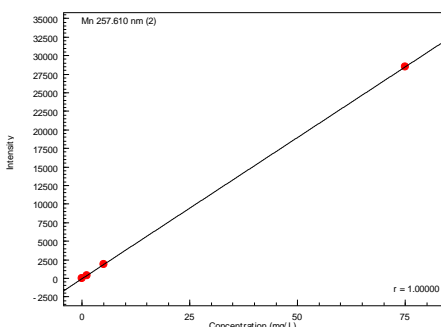
Equation: $\text{Conc.} = a \cdot I^3 + b \cdot I^2 + c \cdot I + d$
 Factor: $a = 0.0000000$ $c = 1.321207e-004$ **Weight: None**
 $b = 2.617437e-010$ $d = -4.133652e-004$ **Offset: None**



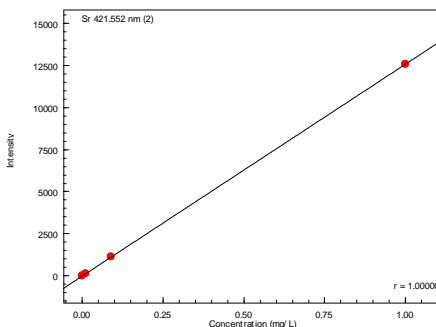
Equation: $\text{Conc.} = a \cdot I^3 + b \cdot I^2 + c \cdot I + d$
 Factor: $a = 0.0000000$ $c = 0.0078761$ **Weight: None**
 $b = 5.197735e-008$ $d = -0.56897163$ **Offset: None**



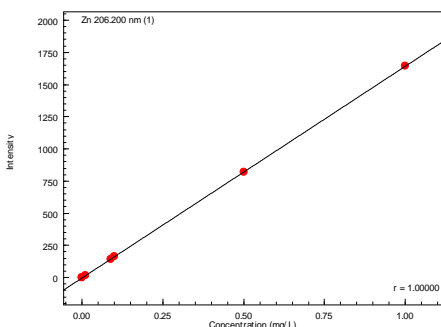
Equation: $\text{Conc.} = a \cdot I^3 + b \cdot I^2 + c \cdot I + d$
 Factor: $a = 0.0000000$ $c = 0.0077268$ **Weight: None**
 $b = 0.0000000$ $d = 2.110572$ **Offset: None**



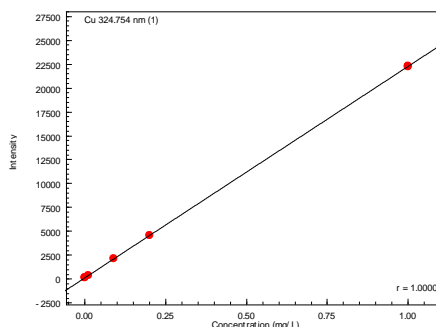
Equation: $\text{Conc.} = a \cdot I^3 + b \cdot I^2 + c \cdot I + d$
 Factor: $a = 0.0000000$ $c = 0.0026289$ **Weight: None**
 $b = 0.0000000$ $d = -0.0135623$ **Offset: None**



Equation: $\text{Conc.} = a \cdot I^3 + b \cdot I^2 + c \cdot I + d$
 Factor: $a = 0.0000000$ $c = 7.937185e-005$ **Weight: None**
 $b = 0.0000000$ $d = -5.427545e-004$ **Offset: None**



Equation: $\text{Conc.} = a \cdot I^3 + b \cdot I^2 + c \cdot I + d$
 Factor: $a = 0.0000000$ $c = 6.081291e-004$ **Weight: None**
 $b = 0.0000000$ $d = -3.684691e-004$ **Offset: None**



Equation: $\text{Conc.} = a \cdot I^3 + b \cdot I^2 + c \cdot I + d$
 Factor: $a = 0.0000000$ $c = 4.515240e-005$ **Weight: None**
 $b = 0.0000000$ $d = -0.0284675$ **Offset: None**

Figure 6: Calibration Curves

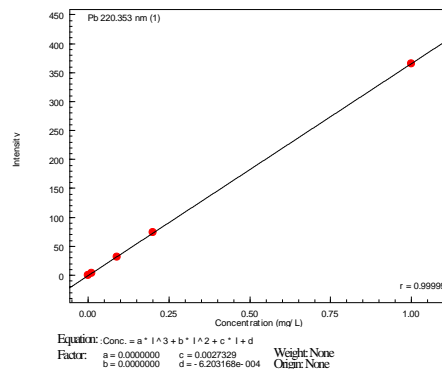
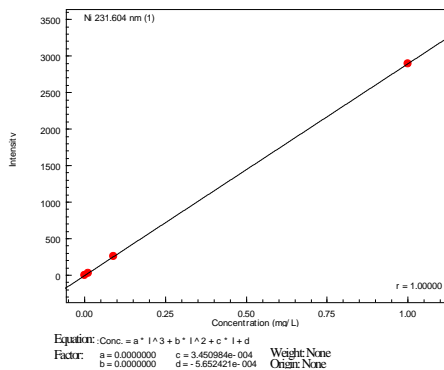
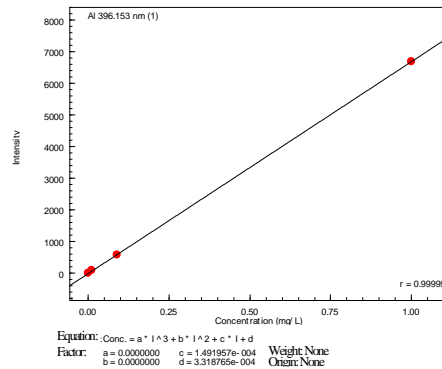
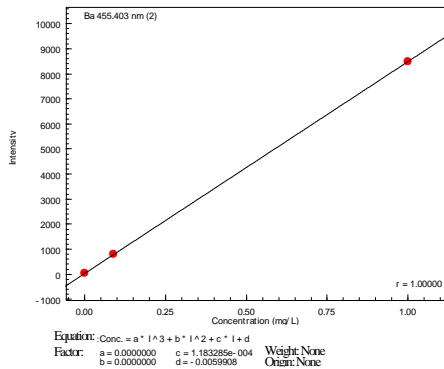
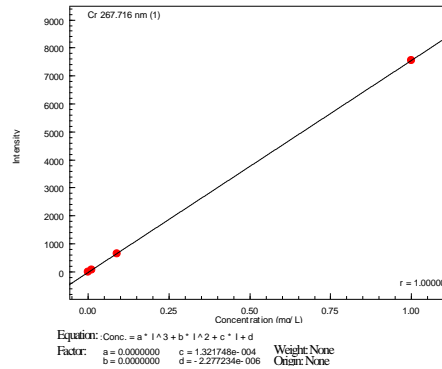
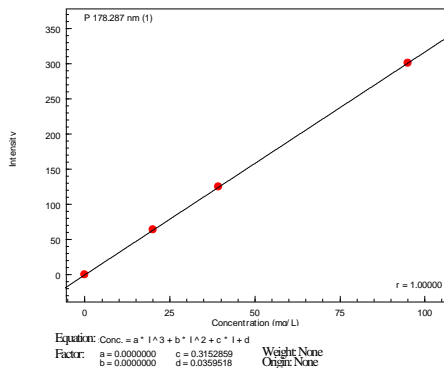


Figure 7: Calibration Curves