

A New Method for the Analysis of Total Nitrogen in Aqueous Samples

Pittcon 2017 2340-5P

William Lipps
Shimadzu Scientific Instruments, Inc.

A New Method for the Analysis of Total Nitrogen in Aqueous Samples

Introduction

The USEPA Clean Water Act (CWA) requires national approval for all methods used for CWA compliance ⁽¹⁾. There are approved methods for the determination of inorganic nitrogen ($\text{NO}_3\text{-N}$, $\text{NO}_2\text{-N}$, and $\text{NH}_3\text{-N}$), and for organic nitrogen (TKN – $\text{NH}_3\text{-N}$). There are no EPA approved methods for the determination of total nitrogen

although it is a required parameter in many USEPA permits, including monitoring of nutrient pollution for ambient water criteria. Because Part 136.3 Table 1b methods are required, laboratories have no recourse but to measure total nitrogen as the sum of TKN, $\text{NO}_3\text{-N}$, and $\text{NO}_2\text{-N}$ ⁽²⁾.

ASTM D8083-16, a New Method for Total Nitrogen

This new method ⁽³⁾ couples the Shimadzu TOC-L (Figure 1) High Temperature Catalytic Oxidation (or Combustion) Total Organic Carbon (TOC) analyzer with the Shimadzu TNM chemiluminescent nitrogen detector. The combustion temperature is 720 °C.

Nitrogen containing compounds in the sample introduced into the combustion tube convert to nitrogen monoxide (reactions 1 and 2). Nitrogen gas in the carrier gas (air) does not interfere. The carrier gas containing the nitrogen monoxide (NO) passes through a thermoelectric cooler. The cooled and dehumidified gas then enters the chemiluminescence analyzer where the NO reacts with

ozone (O_3) and converts to a combination of nitrous oxide (NO_2) and excited nitrous oxide (NO_2^*) (reactions 3 and 4). As the NO_2^* returns to the ground state, it emits radiation that is measured photo-electrically (reaction 5). The detector signal generates a peak that is proportional to the nitrogen concentration in the sample.

- (1) $4\text{NH}_3 + 5\text{O}_2 \rightarrow 4\text{NO} + 6\text{H}_2\text{O}$
- (2) $2(\text{NH}_2)_2\text{CO} + 5\text{O}_2 \rightarrow 4\text{NO} + 4\text{H}_2\text{O} + \text{CO}_2$
- (3) $\text{NO} + \text{O}_3 \rightarrow \text{NO}_2 + \text{O}_2$
- (4) $\text{NO} + \text{O}_3 \rightarrow \text{NO}_2^* + \text{O}_2$
- (5) $\text{NO}_2^* \rightarrow \text{NO}_2 + h\nu$



Figure 1: Shimadzu TOC-L with TN Module

A New Method for the Analysis of Total Nitrogen in Aqueous Samples

Description of the Method

The Shimadzu TOC-L with TN module converts all nitrogen compounds to NO at 720 °C. The instrument uses an auto-sampler to automatically inject sample onto a platinum catalyst inside the heated combustion chamber. The instrument automatically calibrates from a single 10 mg/L N solution to establish a multiple point calibration

curve from 0.2 – 10 mg/L N. The instrument automatically dilutes (or injects less sample aliquot) off-scale peaks, enabling quantitation up to 500 mg/L. Concentrations higher than 500 mg/L N are diluted manually. Total analysis time, per injection, is 2-5 minutes. The Method Detection Limit (MDL) is 0.05 mg/L N.

Comparison with EPA TN Methods

Fifty-two samples were analyzed by the EPA calculation for TN and compared to TN determined by the new ASTM method. These results are statistically the same (Figure 2)

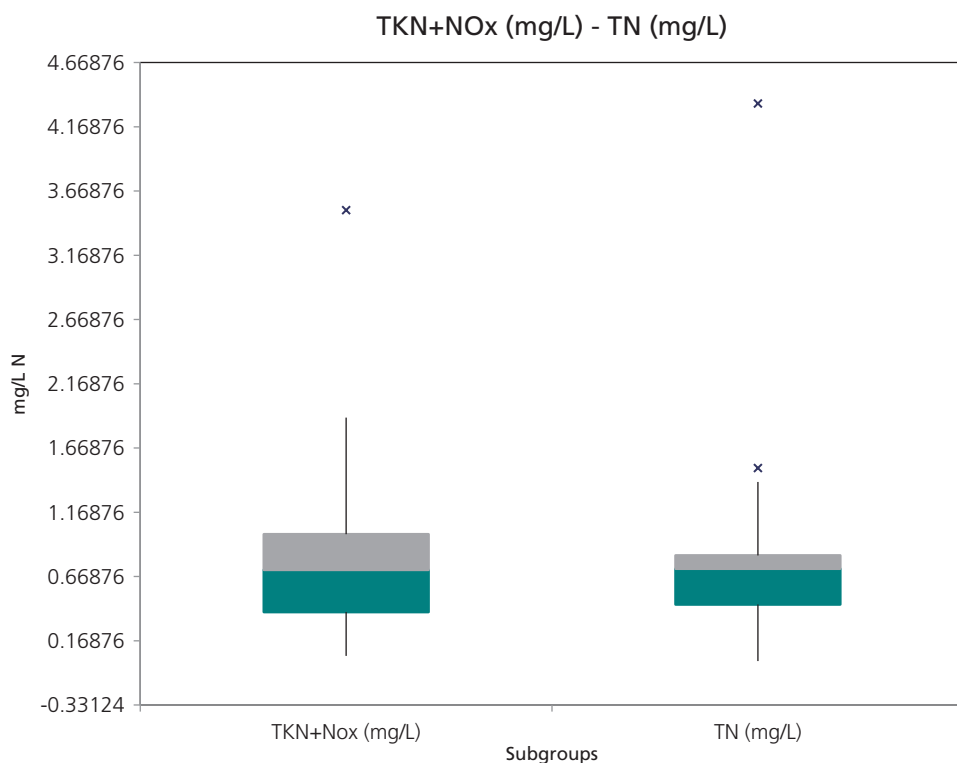


Figure 2: Comparison of EPA TN and ASTM D8083

A New Method for the Analysis of Total Nitrogen in Aqueous Samples

Inter-laboratory Study Data

The ASTM D19.06 task group validating ASTM D8083-16 conducted an inter-laboratory study in accordance with ASTM Practice D2777 – 13 with 8 operators in 7 laboratories using reproducible synthetic matrices, POTW matrices, and a pulp and paper effluent. Laboratories

received each matrix, with the exception of a Laboratory Control Sample (LCS), as blind samples. Table 1 lists the statistical summary for all the inter-laboratory study results.

Table 1: Summary of ILS results

Matrix	Source of N	Known N (mg/L)	Found N (mg/L)	% Recovery	% RSD (multiple operator)	% RSD (single operator)
1	Glycine	5.00	5.15	103	3.56	1.49
		4.00	4.04	101	4.89	0.93
2	Nicotinic Acid	2.00	1.93	97	4.71	1.20
		1.61	1.54	96	4.75	1.06
3	Glycine	0.514	0.496	97	6.03	2.81
		0.313	0.303	97	12.2	4.56
4	Mix of NH ₃ -N and NO ₃ -N	10.0	9.70	97	7.94	1.99
5	NH ₃ -N	30.0	28.5	95	9.50	2.98
6	unknown		29.6		7.74	2.71
7	unknown		4.41		11.5	6.80
8	unknown		9.30		31.0	3.42
9	unknown		339		37.9	6.76

Conclusion

ASTM D8086-16 is a new method for the determination of total nitrogen, and TKN by calculation, in water samples. The method compares favorably with the current EPA approved calculated total nitrogen in samples

without interferences. ASTM D8083-16 saves laboratories time, uses less reagent and smaller sample volumes, and overcomes several known issues with current TKN and TN methods.

A New Method for the Analysis of Total Nitrogen in Aqueous Samples

References

- (1) <https://www.epw.senate.gov/water.pdf>, Secs 301(a), 304 (h), and 501(a), accessed February 10, 2017
- (2) Code of Federal Regulations, Protection of the Environment, Title 40, Chapter 1, Subchapter D, Part 136.3
- (3) ASTM D8083-16, Standard Test Method for Total Nitrogen, and Total Kjeldahl Nitrogen (TKN) by Calculation, in Water by High Temperature Catalytic Combustion and Chemiluminescence Detection, ASTM International, West Conshohocken, PA, 2016, www.astm.org

First Edition: March, 2017



Shimadzu Corporation
www.shimadzu.com/an/

For Research Use Only. Not for use in diagnostic procedures.

This publication may contain references to products that are not available in your country. Please contact us to check the availability of these products in your country.

The content of this publication shall not be reproduced, altered or sold for any commercial purpose without the written approval of Shimadzu. Company names, products/service names and logos used in this publication are trademarks and trade names of Shimadzu Corporation, its subsidiaries or its affiliates, whether or not they are used with trademark symbol "TM" or "®".

Third-party trademarks and trade names may be used in this publication to refer to either the entities or their products/services, whether or not they are used with trademark symbol "TM" or "®".

Shimadzu disclaims any proprietary interest in trademarks and trade names other than its own.

The information contained herein is provided to you "as is" without warranty of any kind including without limitation warranties as to its accuracy or completeness. Shimadzu does not assume any responsibility or liability for any damage, whether direct or indirect, relating to the use of this publication. This publication is based upon the information available to Shimadzu on or before the date of publication, and subject to change without notice.